

# Caltech Chemical Ionization Mass Spectrometer (CIMS) Measurements



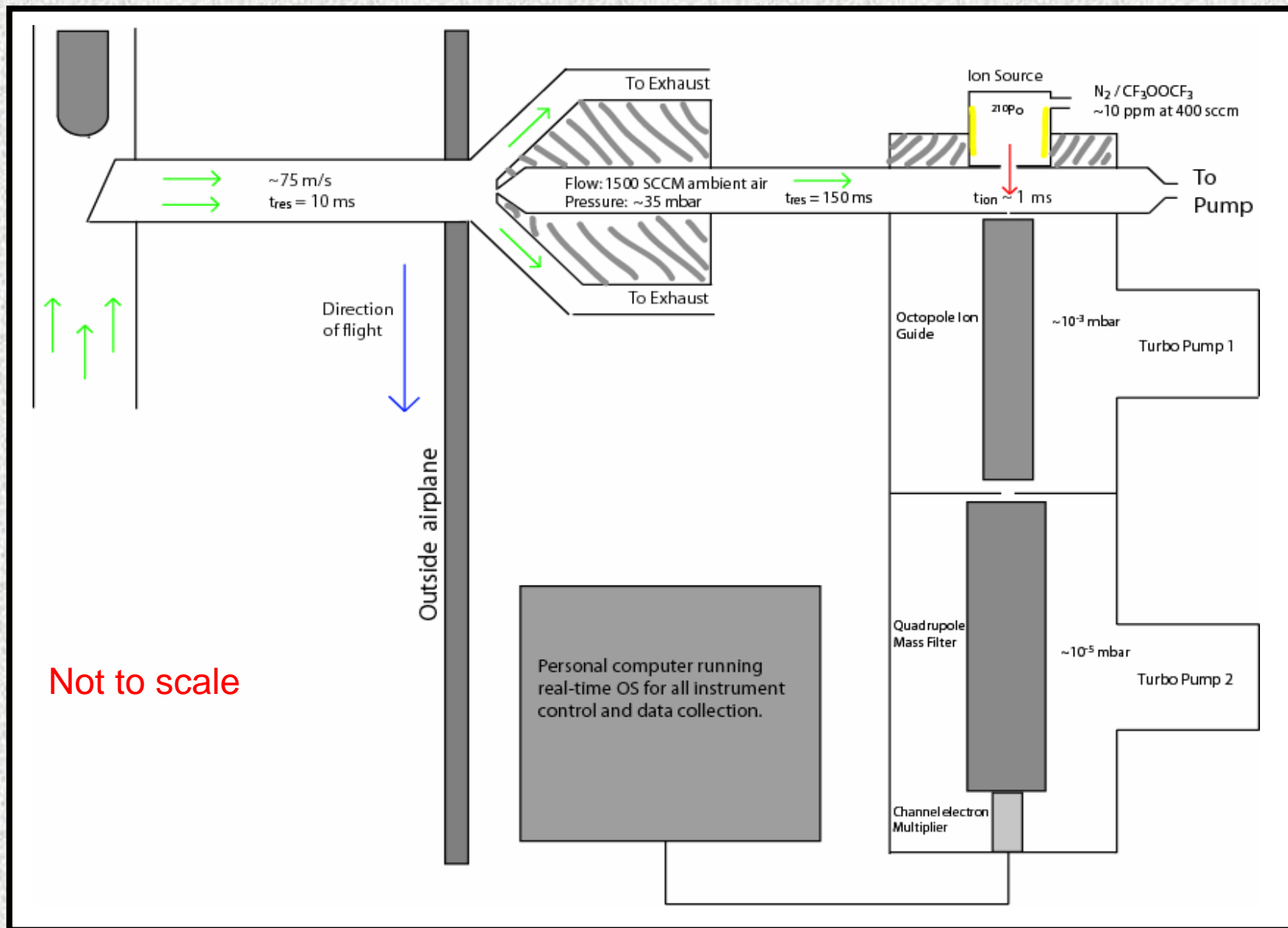
INTEX-NA Data Review

John Crouse

Alan Kwan

Paul Wennberg

# Instrument Block Diagram





# Measurements

- $\text{HNO}_3$  (0.5 s every 5 s)
- $\text{H}_2\text{O}_2$  (0.5 s every 5 s)
- Peroxyacetic Acid (PAA,  $\text{CH}_3\text{C}(\text{O})\text{OOH}$ )
  - (0.5 s every 10 s)
- HCN (Flight #10 onwards)
  - (0.5 s every 10 s)

# Calibration

- Online  $\text{HNO}_3$  calibration from perm tube held at constant temp.
- Online  $\text{H}_2\text{O}_2$  calibration from urea-hydrogen peroxide held at constant temperature.
- Calibrations performed every hour.

# Data Reduction

Concentration =

$$\text{cts\_signal\_ion} / \text{cts\_reagent\_ion} / F([\text{H}_2\text{O}]) * \text{cal\_factor}$$

DLH-H<sub>2</sub>O was used for ambient water concentrations.

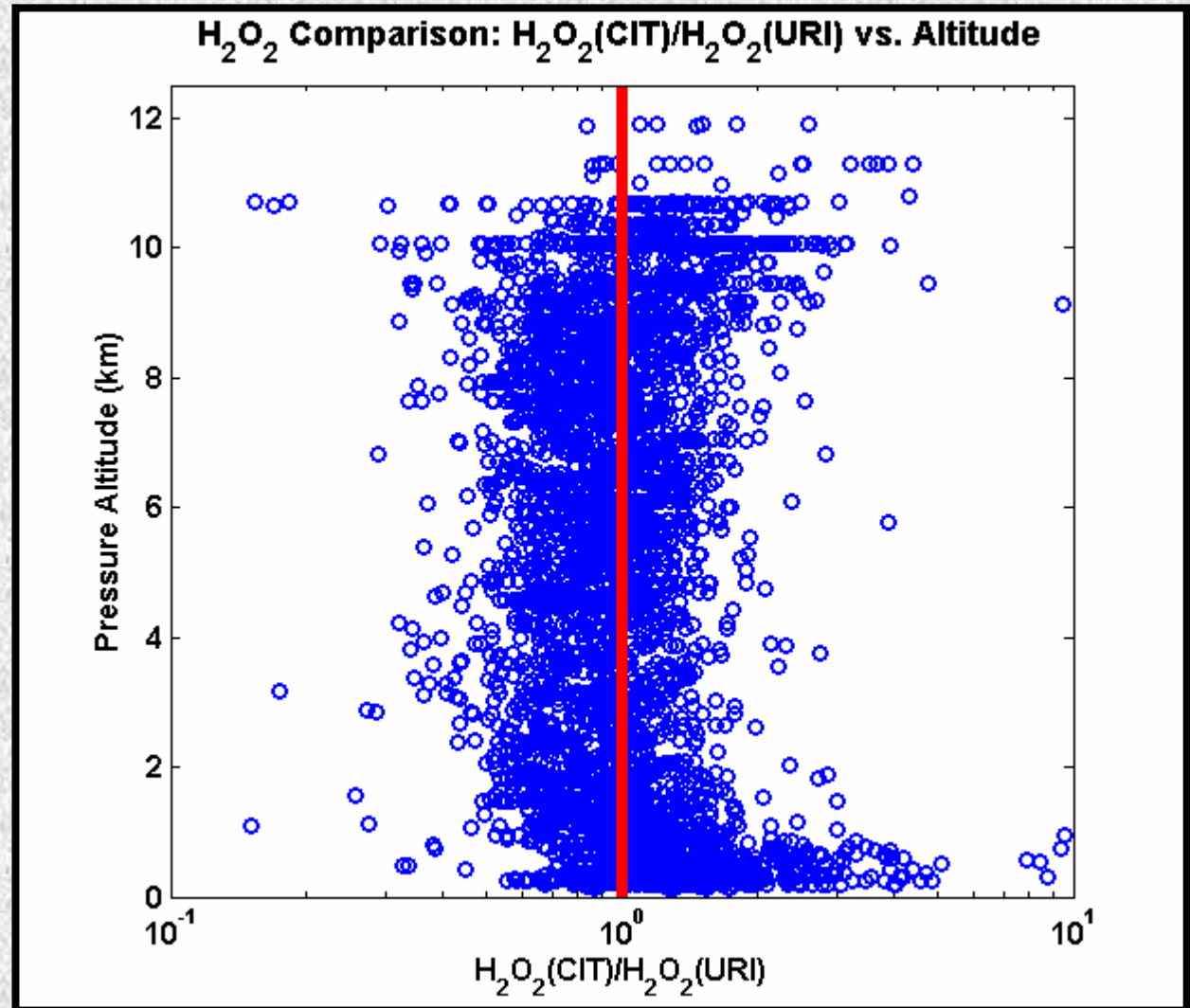
# H<sub>2</sub>O<sub>2</sub> Measurements

- Compare well with URI  
HPLC/Fluorescence measurements
- There is an interference at high SO<sub>2</sub> and high H<sub>2</sub>O, this only affects a small fraction of measurements.

# CIT/URI H<sub>2</sub>O<sub>2</sub> Comparison

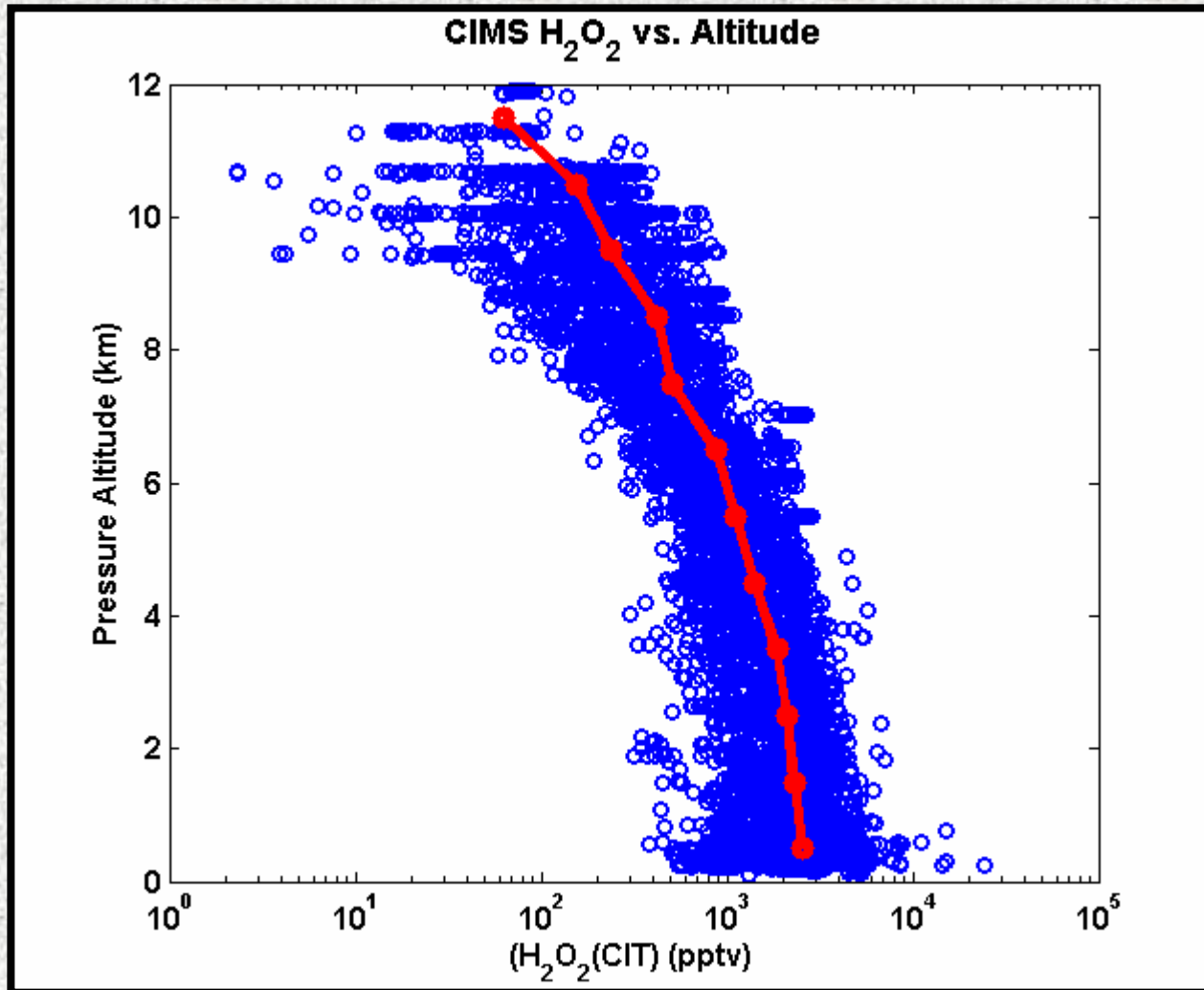
Median: 0.98

Mean: 1.10





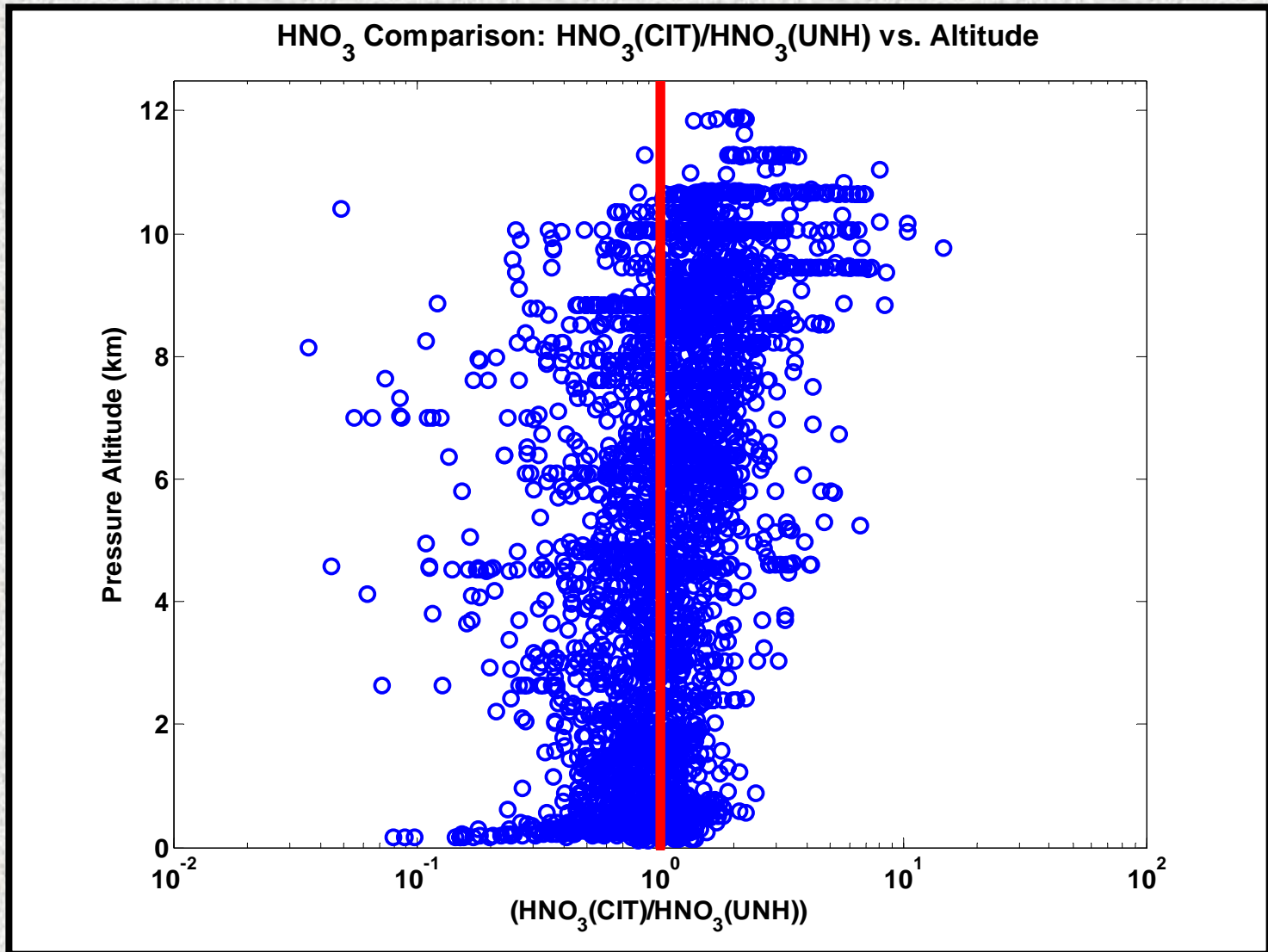
# Altitude Profile $\text{H}_2\text{O}_2$



# HNO<sub>3</sub> Measurements

- Compared with UNH well overall but with certain differences: Altitude trend, low altitude trend, biomass burning plumes.

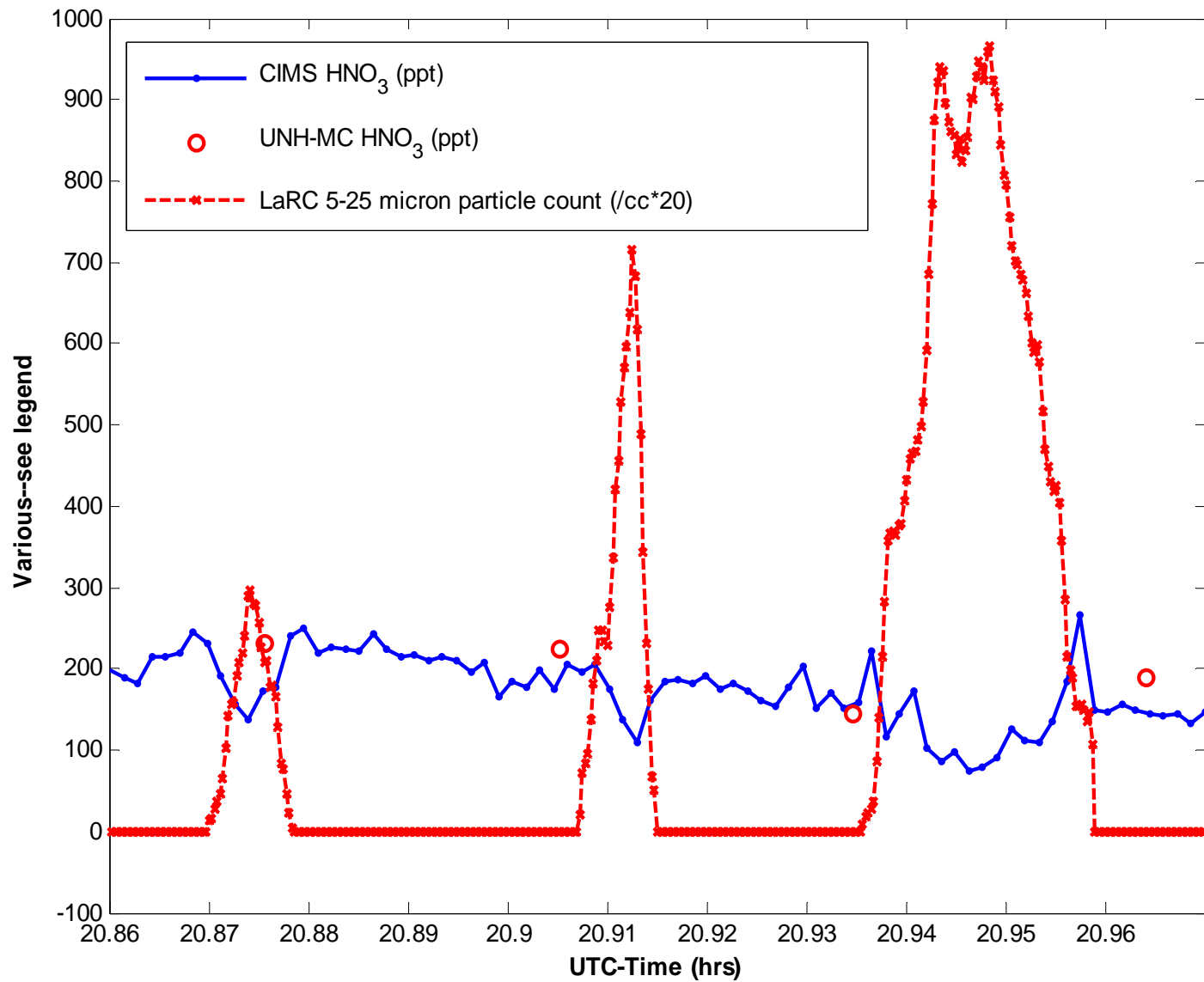
# HNO<sub>3</sub> Comparison



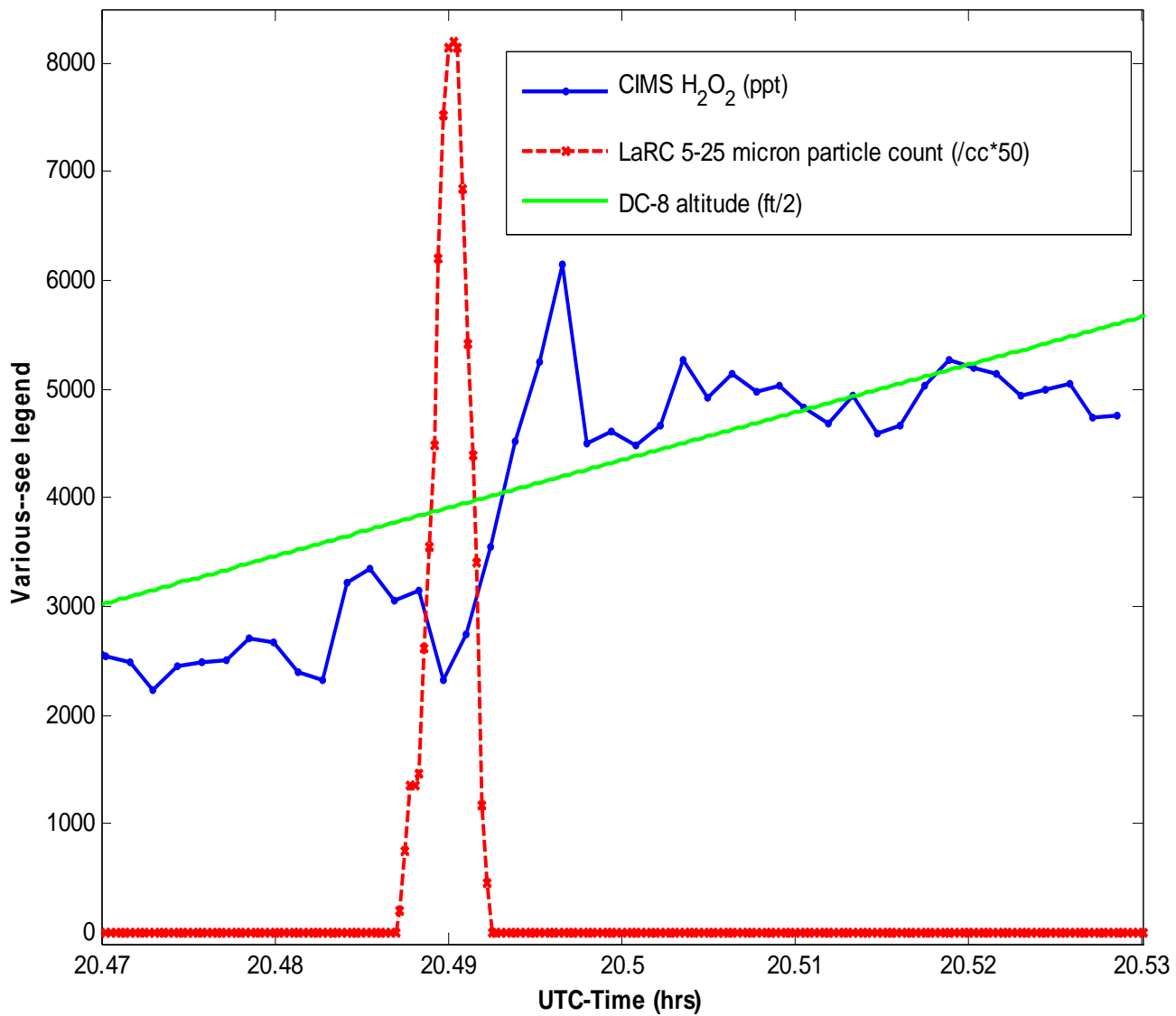
# Fast $\text{HNO}_3$ and $\text{H}_2\text{O}_2$ Measurements

- Interesting data within cloud, and around cloud/clear interfaces.

# HNO<sub>3</sub> Measurements in Clouds, INTEX-NA flight of 040720



# H<sub>2</sub>O<sub>2</sub> Measurements in Clouds, INTEX-NA flight of 040720



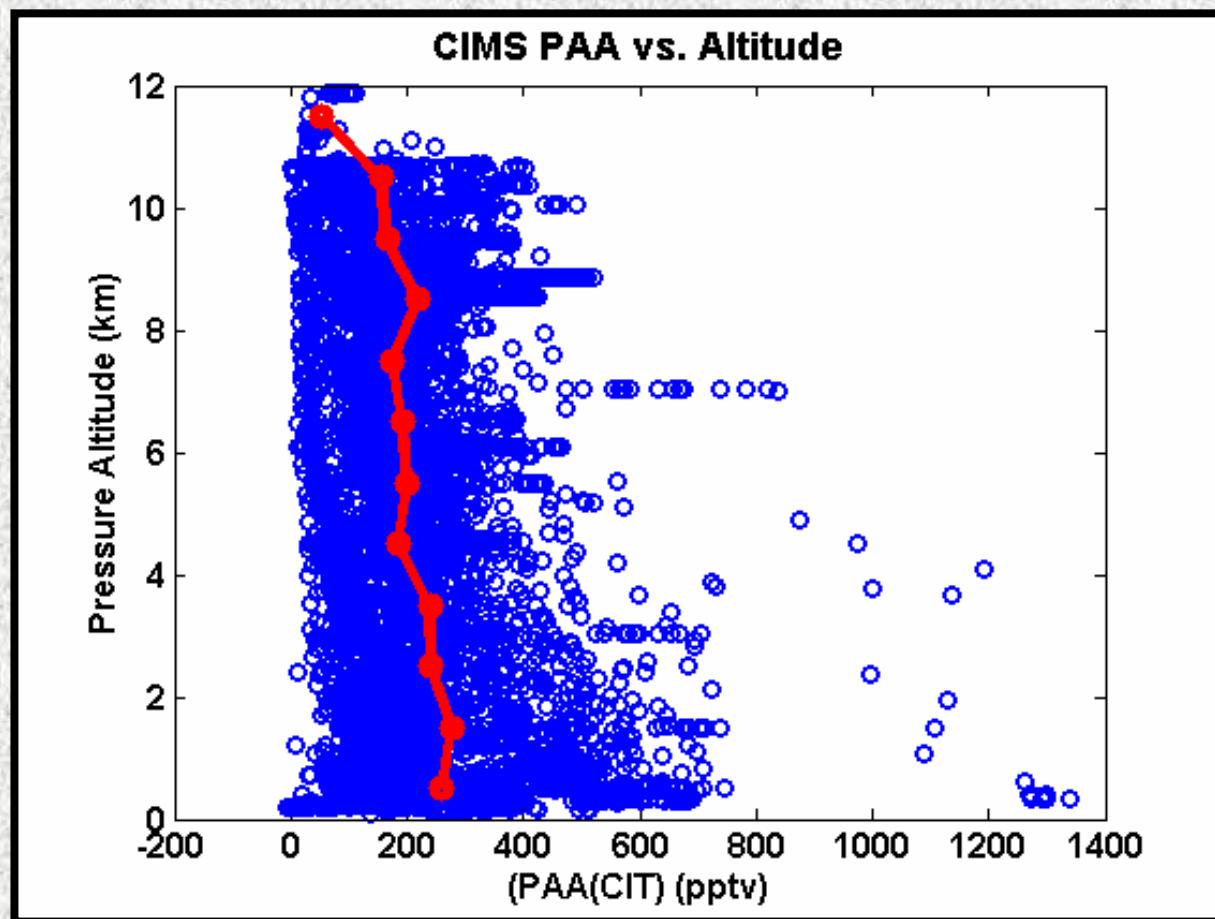
# PAA Measurements

- Measured significant amounts of this molecule throughout mission, though it's abundance was quite variable in space and time.

Significant concentrations observed.  
Mean: 213 pptv  
Median: 188 pptv

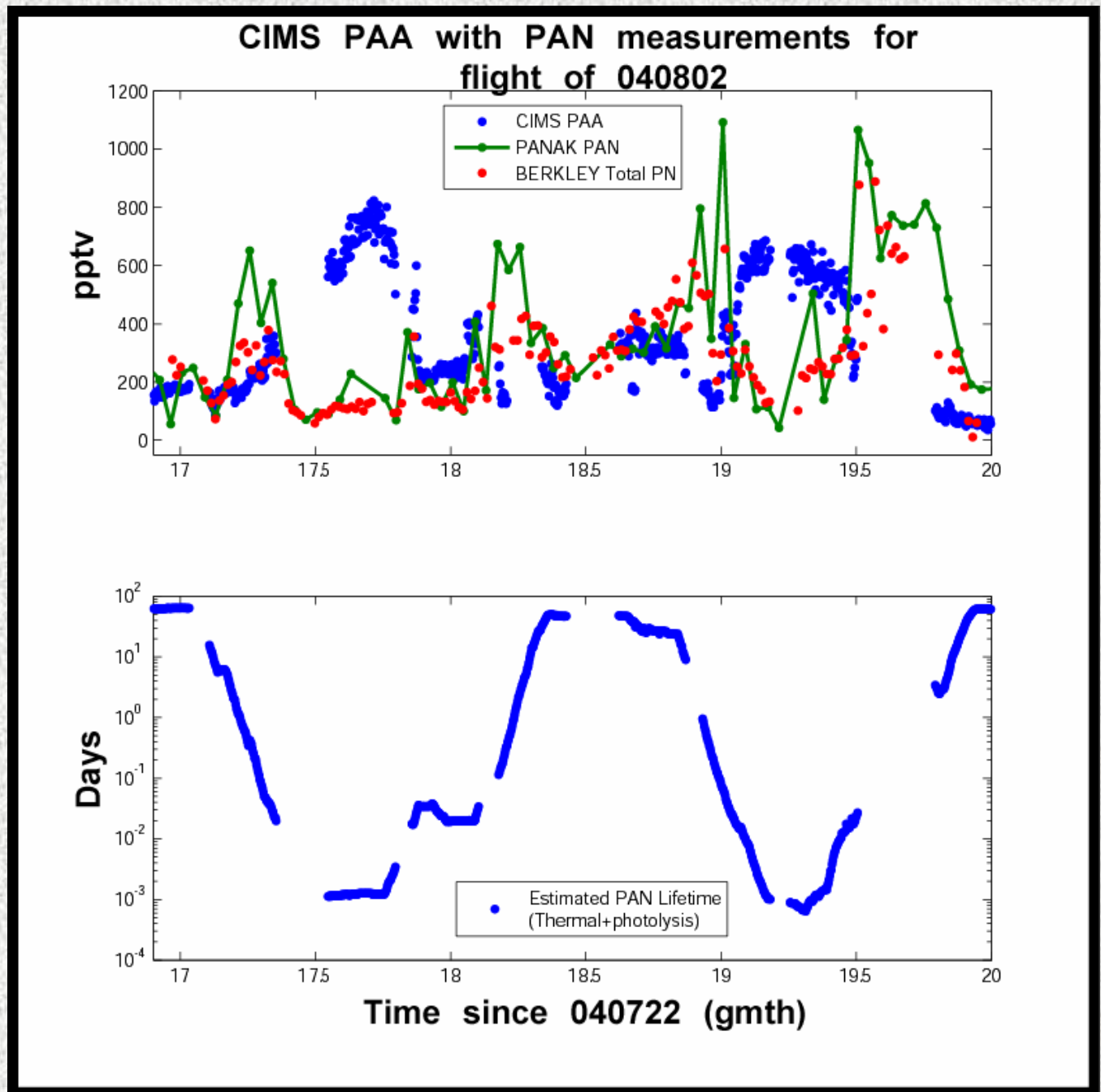
Falls off weakly with altitude.

Highly variable in space and time.





Does PAA come from the thermal decomposition of PAN ?



# PAA Lifetime?

- Atmospheric lifetime??
  - Photolysis: 3-4 weeks  $\rightarrow J = 5 \times 10^{-7} \text{s}^{-1}$   
(Orlando, et. al., 2003)
  - Reaction rate with OH has NOT been measured!!
    - Estimated to be  $1-7 \times 10^{-12}$  from the measured rates of OH with HOOH, and CH<sub>3</sub>OOH.
    - This gives lifetime of 2-12 days.

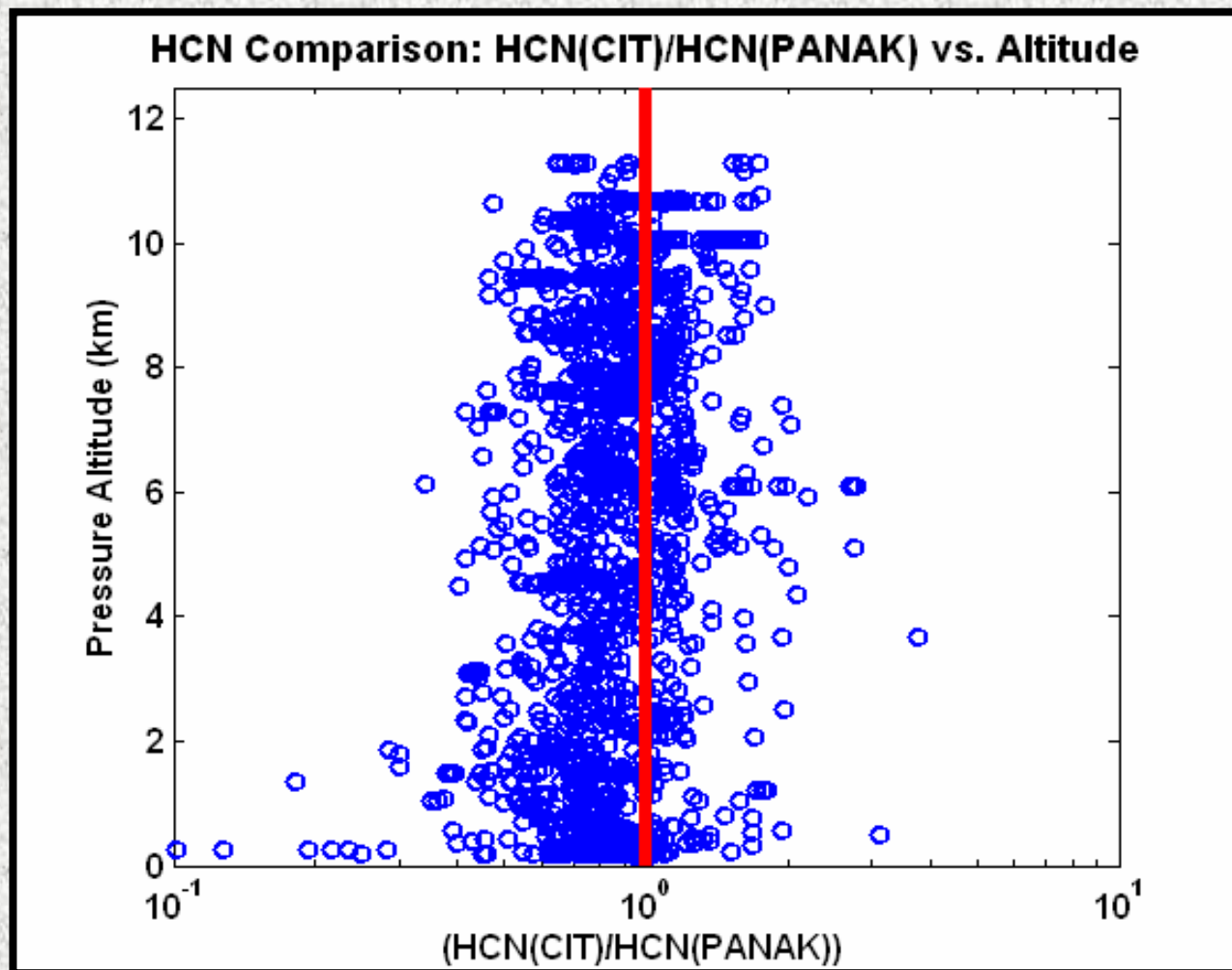
# HCN Measurements

- Fair agreement with PANAK-HCN.
- CIT has high uncertainty at high H<sub>2</sub>O mixing ratios as product ion has a water dependent mass analog interference, as well as decreased sensitivity at high H<sub>2</sub>O.
- Fast HCN measurements may help understand complex air masses which may be a mixture of biomass burning and anthropogenic pollution.

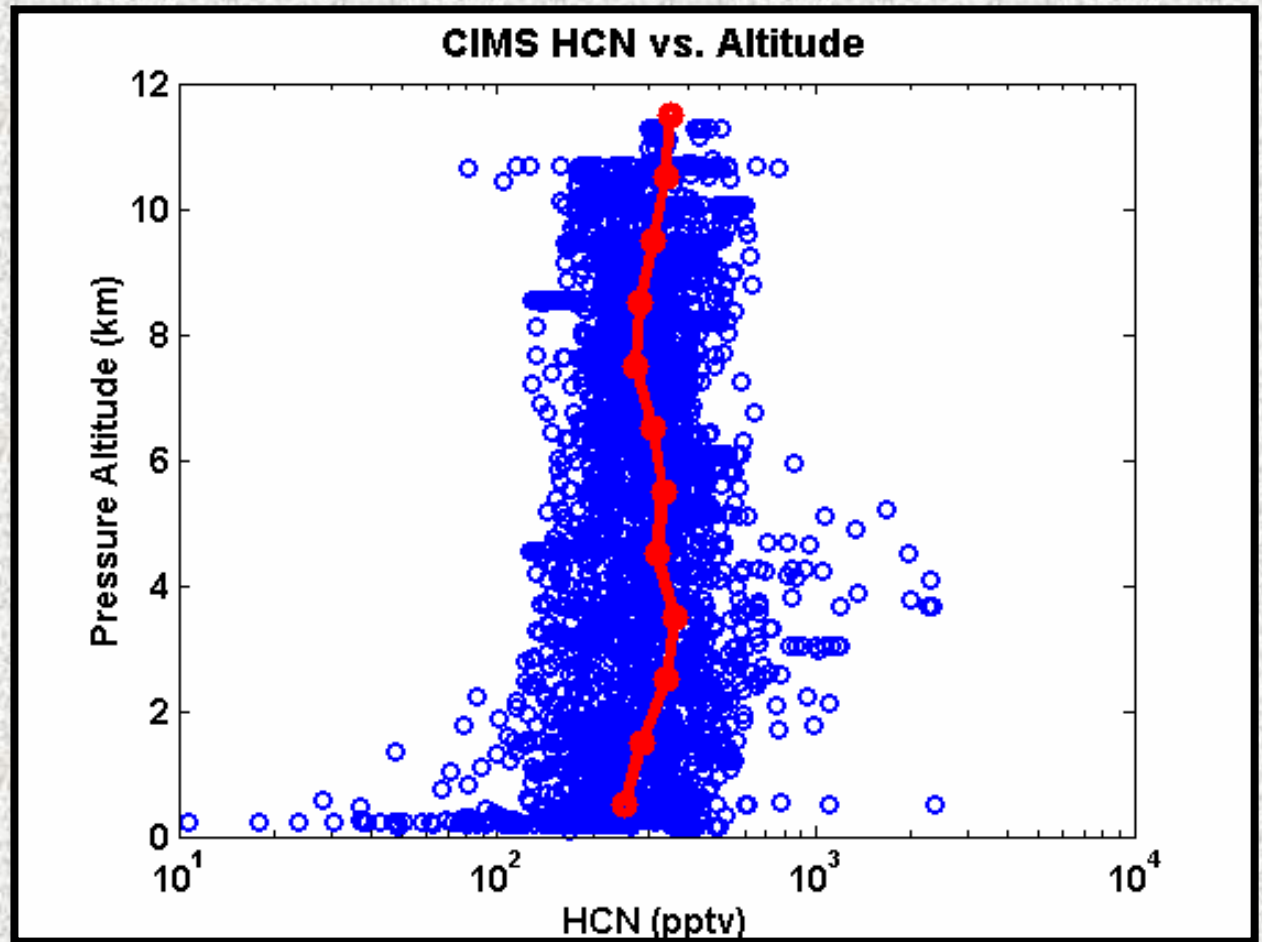
HCN comparison  
versus altitude:

Median: 0.88

Mean: 0.98



HCN altitude profile:  
Median: 279 pptv  
Mean: 302 pptv



# Areas of interest

- Understanding atmospheric importance of PAA: sources, sinks, and lifetime.
  - Utilize model to help constrain these
- Measurements of  $\text{HNO}_3$  and  $\text{H}_2\text{O}_2$  in clouds and around cloud/clear interfaces.
- Use of fast HCN measurements to help constrain ‘complicated’ air masses.